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# **General Principles of Liquid Chromatography**

# Size Exclusion **Chromatography**

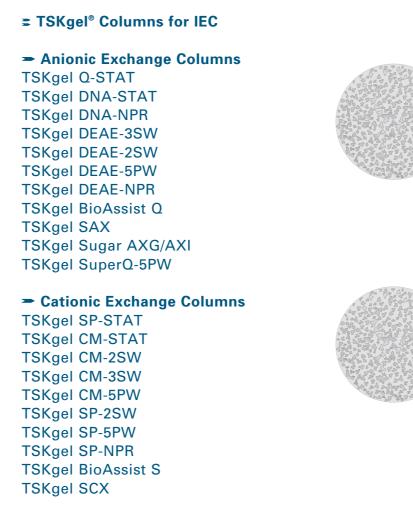
#### **=** TSKgel<sup>®</sup> Columns for SEC

**TSKgel SW-Series** TSKgel SW<sub>XI</sub>-Series TSKgel SuperSW-Series TSKgel UltraSW **TSKgel PW-Series** TSKgel PW<sub>XI</sub>-Series TSKgel SuperMultiporePW TSKgel SuperOligoPW **TSKgel Alpha-Series TSKgel SuperAW-Series TSKgel H-Series** TSKgel H<sub>XI</sub>-Series **TSKgel Super H-Series TSKgel Super HZ-Series** TSKgel SuperMultipore TSKgel High/Ultra-High Temperature GPC

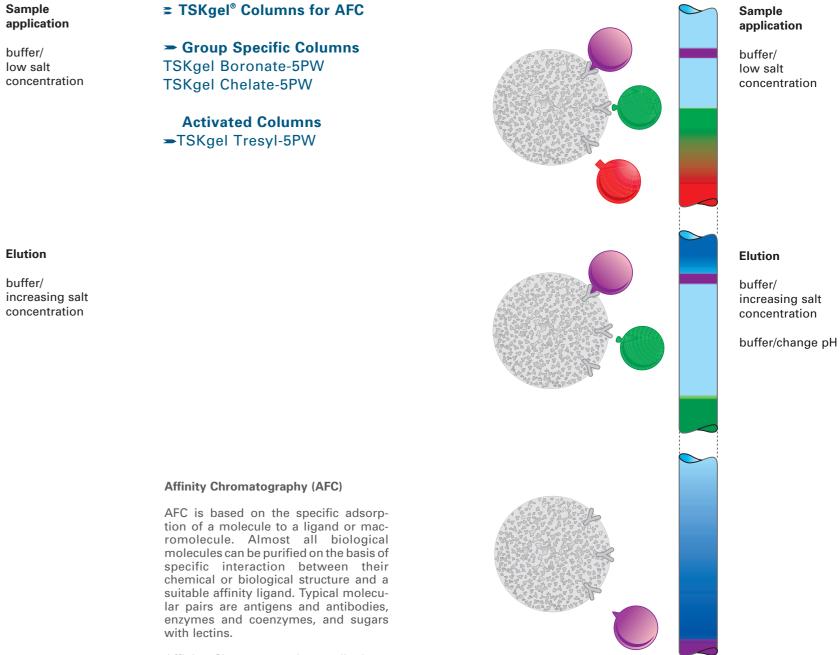
#### Size Exclusion Chromatography (SEC)

is the general name for the chromatographic mode also referred to as gel permeation chromatography (GPC) for non-aqueous elution systems

# Ion Exchange **Chromatography**



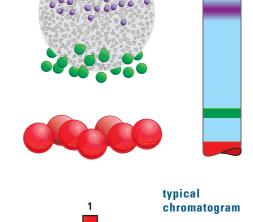
# Affinity **Chromatography**



or gel filtration chromatography (GFC) for aqueous systems.

SEC is a method in which components of a mixture are separated according to their molecular size (hydrodynamic volume), based on the flow of the sample through a porous packing. Large biomolecules that cannot penetrate the pores of the packing material elute first. These large biomolecules are said to be excluded from the packing; they flow with the mobile phase in the interparticle space of the packed column. Smaller molecules can partially or completely enter the stationary phase. Because these smaller molecules have to flow through both, the interparticle space, as well as through the pore volume, they will elute from the column after the excluded sample components

SEC is a very simple method for separating biomolecules, because it is not necessary to change the composition of the mobile phase during elution. However, the separation capacity of this method is limited. For a baseline separation it is necessary that the molecular weights of the biomolecules differ at least 10 to 20 %.



Sample

buffer/

Elution

buffer/

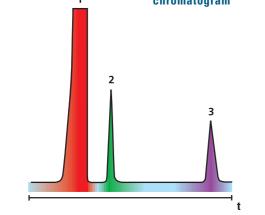
low salt

concentration

low salt

concentration

application

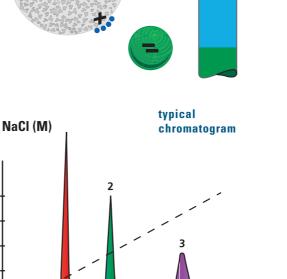


#### Ion Exchange Chromatography (IEC)

Biomolecules generally have charged groups on their surfaces, which change with the pH of the solution. This is the basis for Ion Exchange Chromatography (IEC), in which the molecule reversibly binds to an oppositely charged group of the packing material.

Molecules with a higher charge density bind more strongly to the packing. The bound sample may be selectively removed from the stationary phase by changing the pH or salt concentration of the mobile phase. The higher the charge of the molecule and the stronger the binding to the stationary phase, the greater is the change in the salt concentration required. In IEC it is possible to load samples in a very dilute solution and to elute rapidly with a step gradient, thus producing a concentrated sample IEC is a very powerful separation

tool because it is highly selective and specific and has a high capacity. Although the technique is used for a variety of samples, it is particularly effective for proteins because they are amphoteric. It is estimated that 70 % of all separation methods for proteins involve IEC.



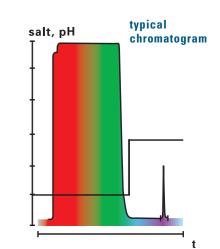
=

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buffer/

Affinity Chromatography media have ligands that are bonded via a spacer arm to the packing material. A specific biological molecule is then reversibly adsorbed to the ligand. The adsorbed molecule is eluted either by competitive displacement or by a change in the conformation of the molecule through a change in pH or ionic strength.

Because of the intrinsic high selectivity of Affinity Chromatography, it is, in contrast to other chromatographic methods, most suitable for specific separation problems and provides high purification yields. Another advantage of AFC is the simplicity of the elution technique, which involves a single-step gradient.



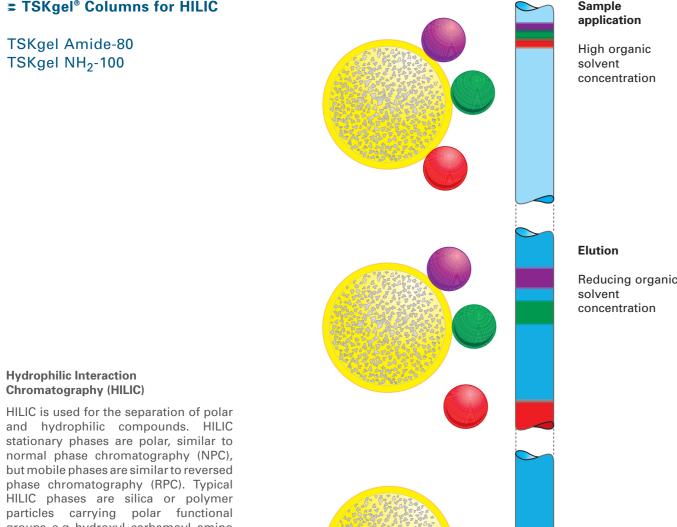
# **Hydrophilic Interaction Chromatography**



**TSKgel Amide-80** TSKgel NH<sub>2</sub>-100

**Hydrophilic Interaction** 

Chromatography (HILIC)



# **Reversed Phase Chromatography**

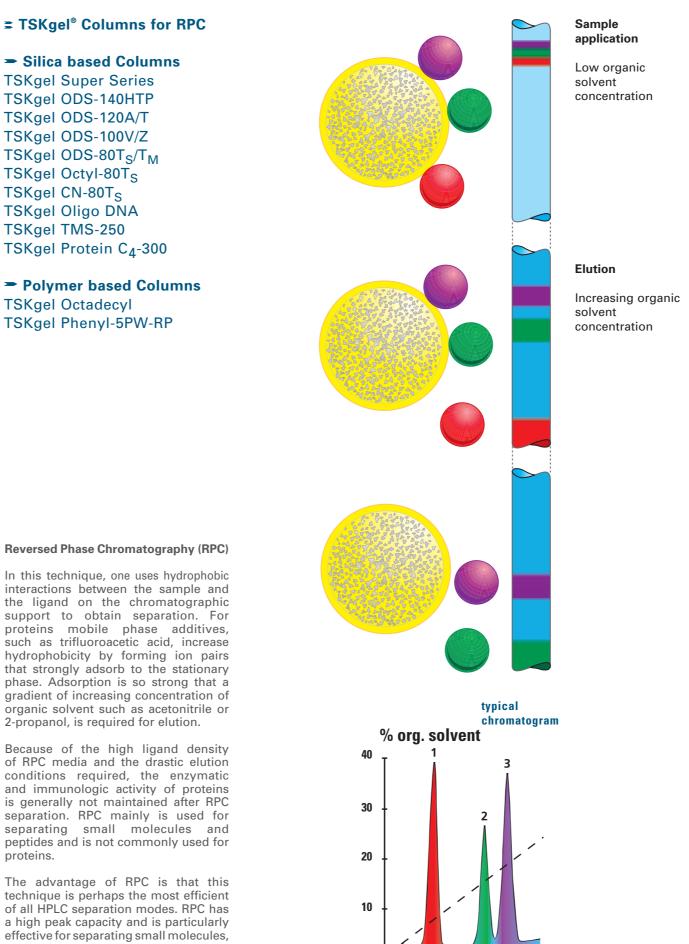
0.5

0.4

0.3

0.2

0.1



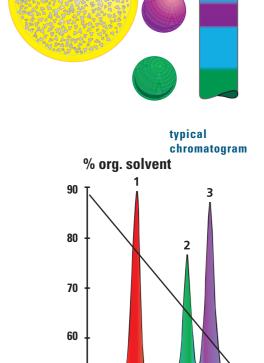
# **Hydrophobic Interaction Chromatography**

#### **=** TSKgel<sup>®</sup> Columns for HIC Sample application **TSKgel Phenyl-5PW** buffer/ **TSKgel Ether-5PW** high salt TSKgel Butyl-NPR concentration Elution buffer/ decreasing salt concentration Hydrophobic Interaction Chromatography (HIC) This is a chromatographic technique in which the sample interacts, at high mobile phase salt concentration, with a hydrophobic stationary phase. Subsequently it is eluted from the

groups, e.g. hydroxyl, carbamoyl, amino or zwitterionic groups. Typical mobile phases are aqueous buffers with organic modifyers - primarily acetonitrile applied in isocratic or gradient mode. In HILIC water has the highest elution power. Therefore gradients start with high percentage of acetonitrile.

It is commonly believed that in HILIC the aqueous content of the mobile phase creates a water rich layer on the surface of the stationary phase. This allows partitioning of solutes between the more organic mobile phase and the aqueous layer. Hydrogen bonding and dipoledipole interactions are supposed to be the dominating retention mechanisms. The number of polar groups, as well as the conformation and solubility of the sample in the mobile phase determine the elution order. Compared to RPC the elution order in HILIC mode is inversed for most compounds.

HILIC is ideally suited for mass spectrometric analysis of water soluble polar compounds, because the high organic content in the mobile phase increases MS detection sensitivity.



In this technique, one uses hydrophobic interactions between the sample and the ligand on the chromatographic support to obtain separation. For proteins mobile phase additives, such as trifluoroacetic acid, increase hydrophobicity by forming ion pairs that strongly adsorb to the stationary phase. Adsorption is so strong that a gradient of increasing concentration of organic solvent such as acetonitrile or 2-propanol, is required for elution.

Because of the high ligand density of RPC media and the drastic elution conditions required, the enzymatic and immunologic activity of proteins is generally not maintained after RPC separation. RPC mainly is used for separating small molecules and peptides and is not commonly used for proteins.

technique is perhaps the most efficient of all HPLC separation modes. RPC has a high peak capacity and is particularly effective for separating small molecules, peptides, nucleotides, and restriction fragments.

increasing the salt concentration of the solvent, these hydrophobic patches of the molecule become more exposed and can interact with hydrophobic ligands on the HIC packing. HIC is particularly attractive for protein purification when the sample is solved in high salt concentration

stationary phase by decreasing the salt

concentration. Almost all biological

molecules have in their structure

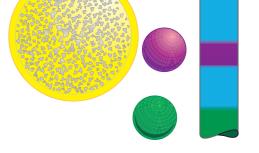
hydrophobic patches that, under

physiological conditions, are shielded

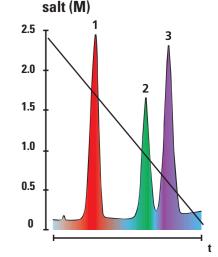
by hydrophilic or ionic groups. By

In contrast to the conditions used in RPC, the biological activity of the eluted molecules is often maintained in HIC. It is being used increasingly as a substitute for ammonium sulfate precipitation because of higher throughput and greater recovery of enzymatic activity.

The strength of the hydrophobic interaction is influenced strongly by the nature of the salt components in the mobile phase. Starting salt concentration of 1.0 M to 2.5 M ammonium sulfate in the buffer is commonly used to adsorb the sample to the packing. The salt concentration needed depends on the protein hydrophobicity and solubility. the resins hydrophobicity and the resolution, capacity and mass recovery required. Additives commonly used are methanol, ethanol, isopropanol, acetone, SDS, urea and guanidinium hydrochloride.







#### TOSOH BIOSCIENCE'S BIOSEPARATION COLUMNS

The analysis, isolation, and purification of biomolecules can be accomplished by a number of chromatographic modes. Each mode is based on specific physical, chemical, or biological interactions between the sample biomolecules and the packing material.

The various modes of chromatography involve separations that are based on specific features of the target or sample, like size, charge, hydrophobicity, function or specific content of the molecule. The general principles of the most commonly used modes are outlined here.

### TOSOH BIOSCIENCE offers a comprehensive line of media and prepacked columns for all common modes of liquid chromatography including ion-exchange, hydrophobic and hydrophilic interaction, reversed phase, size exclusion and affinity. TSKgel is available as bulk polymeric resin or in silica or polymeric-based prepacked columns.

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